

# PATENT SPECIFICATION

(11) 1250719

1250719

## NO DRAWINGS

- (21) Application No. 14529/69 (22) Filed 19 March 1969  
 (23) Complete Specification filed 16 March 1970  
 (45) Complete Specification published 20 Oct. 1971  
 (51) International Classification C 07 d 29/28 29/20 99/02 57/00  
 A 61 k 27/00/ C 07 d 29/10 C 07 c 59/20 149/40  
 (52) Index at acceptance

C2C. 172—194—284 173—197—288 215 220 221 225 227  
 22Y 246 247 250 251 25Y 28X 29X 29Y 304  
 30Y 323 32Y 351 355 364 366 367 36Y 373 37Y  
 3A12A4A 3A12A4C 3A12B3 3A12C1 3A12C6  
 3A14A3D 3A14A5 3A14A8A 3A7V3A2 3A7V3E1  
 3A7V3K1 456 45Y 463 464 552 553 612 613 61X  
 620 625 628 660 662 665 668 675 770 771 790  
 79Y LM NK QT QU ZD

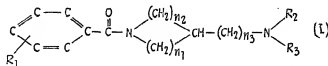
- (72) Inventors OLE BENT TVAERMØSE-NIELSEN, HANS-HASSO  
 FREY and PETER WERNER FEIT

## (54) PIPERIDINE COMPOUNDS FOR THE TREATMENT OF PARKINSONISM

- (71) We, LØVENS KEMISKE FABRIK  
 PRODUKTIONS-SKATIESELSKAB, a Company, in-  
 corporated under the Laws of Denmark, of  
 Ballerup, Denmark, do hereby declare the in-  
 vention, for which we pray that a patent may  
 be granted to us, and the method by which

it is to be performed, to be particularly  
 described in and by the following state-  
 ment:—

This invention relates to a series of hither-  
 to unknown compounds of the general for-  
 mula:



- wherein  $R_1$  is a straight or branched C3 to  
 C12 aliphatic hydrocarbon chain, unsubstituted  
 or substituted with phenyl, phenoxy  
 or phenylthio, which  $R_1$  is directly attached  
 to the benzene nucleus, or optionally may be  
 attached to the benzene nucleus through a  
 hetero atom which is an oxygen or sulphur  
 atom;  $R_2$  is alkyl,  $R_3$  is a cycloalkyl radical  
 with from 5 to 8 carbon atoms in the ring,  
 and  $R_2$  together with  $R_3$  and the nitrogen atom  
 can complete a heterocyclic ring which may  
 be alkyl-substituted,  $n_1$  is an integer from  
 2 to 4,  $n_2$  is an integer from 1 to 5, and  
 $n_3$  is an integer from 1 to 3; to salts of the  
 compounds with pharmaceutically acceptable  
 inorganic and organic acids; and to methods  
 for the preparation of the compounds and  
 their salts.

- When ever used in the statement above  
 or in the description below, the term "alkyl"  
 means lower-alkyl, including straight and  
 branched aliphatic hydrocarbon chains with  
 from 1 to 6 carbon atoms in the chain.

- The compounds of the invention possess

valuable pharmacological activities, thus e.g.  
 they display a favourable central anticholin-  
 ergic action and are intended to be used in  
 the treatment of patients suffering for in-  
 stance from parkinsonism, including the post-  
 encephalytic or arteriosclerotic parkinsonism  
 and similar conditions.

As implied in the term, post-encephalytic  
 parkinsonism refers to the appearance as a  
 sequence to encephalitis of muscle rigidity  
 and tremor frequently along with spasmodic  
 phenomena, whereas the term arteriosclerotic  
 parkinsonism refers to the appearance as a  
 consequence of multiple cerebral vascular le-  
 sions of difficulties of movements and fixity  
 of posture, and similar conditions occurring  
 in the older age group, often combined with  
 muscle rigidity while tremor is absent. The  
 said disorders are chronic and progressive and  
 consequently all treatment is symptomatic  
 and must be continued for long periods of  
 time.

The medication may comprise treatment  
 with belladonna alkaloids, e.g. atropine,

amphetamine alone or in combination with belladonna alkaloids, with certain antihistaminics or apomorphine, and similar unspecific medications, which may offer some degree of symptomatic relief on tremor or spasmodic conditions, but no fixed dosage can be recommended and ordinarily small amounts of the drug in question are used initially while larger doses are ultimately required whereby it may be necessary to approach the limit of tolerance and several toxic symptoms appear. Better results in the treatment of parkinsonism have been observed by using certain synthetic drugs as e.g. trihexaphenidyl (3 - (1 - piperidyl) - 1 - phenyl - 1 - cyclohexyl - 1 - propanol), Caramiphen (2 - diethylaminoethyl - 1 - phenyl - cyclopentane - 1 - carboxylate hydrochloride), or Diethazide (diethylaminoethyl - N - dibenzoparathiazine).

The action of trihexaphenidyl resembles that of atropine, in particular as far as the antispasmodic properties are concerned whereas some of the undesired effects of atropine are weaker, but still the peripheral parasympatholytic action of trihexaphenidyl must be considered an undesired effect in the treatment of parkinsonism where in particular the central action is important.

As far as the chemical constitution is concerned the compounds of the present invention differ far from the drugs mentioned above, and it has surprisingly been found that the compounds of formula I exert a favourable specific therapeutic action with a view to the treatment of all forms of parkinsonism.

According to experiments the preferred compounds with a view to treatment of parkinsonism are those of formula I in which  $R_1$  is a C5 to C7 aliphatic hydrocarbon chain attached to the benzene nucleus through the hetero atoms O or S, in which the integers  $n_1$  and  $n_2$  are within the limits from 2 to 3, and from 2 to 4 respectively, and in which  $R_2$  is a C1 to C2 aliphatic alkyl group, and  $R_3$  is a C4 to C7 cycloalkyl group, or in which  $R_2$  and  $R_3$  together with the N atom form a heterocyclic ring.

In particular, however, the preferred compounds are those in which  $R_1$  has the meaning defined above and are in the 4-position in the benzene nucleus, and in which  $R_2$  and  $R_3$  together form an unsubstituted or alkyl-substituted pyrrolidino, piperidino, hexamethylenecimino or heptamethylenecimino group.

Thus the compound 1 - (4 - n - hexyloxybenzoyl) - 4 - (piperidinoethyl) - piperidine hydrochloride, among a series of related compounds, displayed a promising central anticholinergic activity, while its peripheral parasympatholytic effects were comparably weak. Its antagonistic effects against the tremorogenic action of tremorine and oxotremorine, which is considered to be the most

predictive pharmacological model of parkinsonism, were two to five times stronger than those of trihexaphenidyl bearing at present the drug of choice in the treatment of parkinsonism. Furthermore, the central effects of oxotremorine (tremor) were inhibited with lower doses than the peripheral effects (salivation) which as mentioned above is highly desirable for antiparkinsonism drugs.

Experiments in higher animals further confirmed the favourable weak peripheral anticholinergic action of the compounds of the invention.

The acute oral toxicity of e.g. 1 - (4 - n - hexyloxybenzoyl) - 4 - (piperidinoethyl) - piperidine HCl expressed in LD<sub>50</sub> (mice) is 470 mg/kg. which may be considered low when compared to the degree of activity in the anti-parkinson test in which an effect could be observed with amounts of the order 0.5 to 2.0 mg/kg.

The chronic toxicity was studied in animal experiments in which the test animals were rats (Leo Wistar strain). The compounds were administered orally each day in a period of six months in various doses, in one animal section in a daily dose of 50 mg/kg.

Even in this latter section no adverse clinical signs were seen and no adverse changes in bodyweight could be demonstrated. The investigation comprises a full haematological and pharmacological analysis of the animals and after post-mortem examinations no abnormalities were demonstrated.

A pharmaceutical composition containing a compound of the invention also constitutes part of this invention. In the composition, the proportion of therapeutically active material to carrier substances and auxiliary agents can vary between 0.04 to 10% depending upon the form of pharmaceutical presentation.

The composition in question can be worked up to pharmaceutical forms of presentations such as tablets, pills, drages and suppositories, or the composition can be filled in medical containers such as capsules or ampoules or, as far as mixtures or elixirs are concerned, they may be filled in bottles and similar containers.

Pharmaceutical inorganic or organic, solid or liquid carriers suitable for enteral and parenteral administration can be used to make up the composition; water, gelatine, lactose, starch, magnesium stearate, talc, vegetable and animal oils and fats, benzyl alcohol, gum, polyalkylene glycol and similar other known carriers for medicaments are suitable as carriers while stabilizing agents, wetting or emulsifying agents, salts for varying the osmotic pressure or buffers for securing an adequate pH-value of the composition can be used as auxiliary agents.

In the composition, the compounds of for-

mula I may be present as such or in the form of one of their salts with a pharmaceutically acceptable inorganic or organic acid as for instance a hydrochloride, hydrobromide, hydriodide, sulphonate, sulphate, phosphate, sulphamate, tartrate, malate, citrate, acetate, succinate or benzoate.

Another object of the invention resides in the selection of a dose of the compounds of formula I which can be administered so that the desired activity is achieved without simultaneous secondary effects.

The compounds of the invention may in clinical practice conveniently be administered by injection, preferably once per day and in amounts corresponding to a total daily dose of from 0.1 to 25 mg.

In particular, however, the compounds may be given by the oral route in the form of tablets or capsules, or in the form of a mixture or elixir, one to four times per day and in a total daily dose of from 0.2 to 50 mg., always with due regard to the condition of the patient and in accordance with the prescription of the medical practitioner.

The compounds of the invention may conveniently be administered in dosage units containing not less than 0.5 mg., and preferably from 1 to 25 mg. of the active compound.

By the term "dosage unit" is meant a unitary, i.e. a single dose capable of being administered to the patients, and which may be readily handled and packed remaining as a physically stable unit dose comprising either the active material as such or as a mixture of it with solid or liquid pharmaceutical diluents or carriers.

If the compound is to be injected, a sealed ampoule, a vial or a similar container may be provided containing a parenterally acceptable aqueous or oily injectable solution or dispersion of the active material as a dosage unit mentioned above.

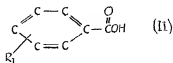
When in the form of tablets, pills or capsules, the dosage unit may contain from 0.5 to 25 mg. and preferably contains from 1 to 10 mg. of the active compound which is readily absorbed when orally administered.

When in the form of an injectable preparation the dosage unit preferably contains from 0.1 to 25 mg. of the active compound.

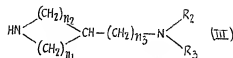
When the active compound is administered as a mixture or elixir, this pharmaceutical preparation may preferably contain 0.5 to 10 mg. per cc. The dosage units aforesaid also constitute part of the present invention.

It is still another object of the invention to provide a method of producing the compounds represented by the general formula I.

In the method of the invention, an acid of the general formula:



wherein  $\text{R}_1$  has the meaning hereinbefore defined is reacted in the form of an acid halide, on anhydride, a mixed anhydride with an alkyl-carbonic acid, a carboxylic acid, a sulphonic acid or with an inorganic acid, or in the form of a reaction derivative obtained by reacting the acid with a carbodiimide or  $\text{N},\text{N}'$ -carboxyldiimidazole, with a compound of the general formula:



wherein  $\text{R}_2$ ,  $\text{R}_3$ ,  $n_1$ ,  $n_2$  and  $n_3$  have the meanings hereinbefore defined, whereafter the compound formed is recovered as such or as one of its salts with an acid.

Most of the starting substances of formulae (II) and (III) are known compounds the preparation of which is described in the literature, or they may, if not known, be prepared in analogy with the known compounds.

Thus e.g. certain functional reactive derivatives of the alkylthiobenzoic acids used as starting substances in the method are hitherto unknown compounds which may be prepared in a Sandmeyer reaction in which the diazotized *m*- or *p*-aminobenzoic acid is reacted with an alkalimetal xanthogenate to form the corresponding xanthate, which in one step is hydrolyzed and alkylated by reacting with an alkylhalide under alkaline conditions resulting in the desired alkylthiobenzoic acid.

As indicated the acid of formula (II) is used in the form of one of its reactive derivatives as for instance an acid halide, such as an acid chloride or bromide, an anhydride, a mixed anhydride with an alkyl-carbonic acid, such as isobutyl-carbonic acid, a carboxylic acid, an inorganic acid or a sulphonic acid; or a derivative obtained by reacting the corresponding free acid with a carbodiimide or  $\text{N},\text{N}'$ -carboxyldiimidazole.

The process of the invention is conveniently performed in the presence of an inert solvent and in the absence of water at room temperature for a period of time necessary to accomplish the desired degree of conversion, commonly by standing overnight. In this embodiment equimolar amounts or an excess

of the compound of formula (II) may appropriately be employed in order to form e.g. the hydrogen halide of compounds of formula (I) directly in the reaction mixture.

5 In another embodiment the reaction is performed in the presence of an inert solvent, preferably immiscible with water, and at temperatures at or below 0° C. while the acid component possibly formed during the reaction e.g. a hydrogen halide, is continuously removed by adding an aqueous solution of a base, e.g. an alkali metal hydroxide. In this embodiment the starting substances are used in equivalent amounts, or substantially in equivalent amounts, and the reaction may be completed within a few hours.

After complete reaction, the desired compound is readily recovered from the organic phase, if necessary after having removed a possible excess of starting substance of formula (II) by extraction with an aqueous solution of an inorganic base, by evaporation of the organic phase, and recrystallizing the residue, or the compound may be isolated as a salt with an acid by neutralizing the base, in a suitable solvent or mixture of solvents with a view to the precipitation or crystallization of the salt.

The invention will now be illustrated by the following non-limiting Examples, of which Examples 1 to 4 illustrate the preparation of intermediates and Examples 5 to 8, illustrate the preparation of the compounds of formula (I):—

#### EXAMPLE 1

##### 4-n-Hexylthiobenzoic acid

To a solution of 4 - aminobenzoic acid (32 g.), sodium nitrite (18.8 g.) and sodium hydroxide (11 g.) in water (150 ml.) was slowly added while stirring rigorously at -5—0° C. After the addition was completed the stirring was continued for a further 1 hour at 0—5° C. The cooled diazonium-solution was filtered and slowly added to a solution of potassium xanthogenate (62.5 g.) and sodium carbonate (87.5 g.) in water (250 ml.) while stirring vigorously at 65—70° C. The mixture was stirred at 65—70° C. for a further 1 hour. After cooling the mixture was carefully acidified with concentrated hydrochloric acid (150 ml.). The precipitated material was filtered off, washed with water and dissolved in 10% sodium hydroxide solution (500 ml.). The flask was filled with nitrogen, closed and left overnight. n-Hexylbromide (85 g.) was added and the mixture was refluxed for 3 hours. The resulting mixture was poured into concentrated hydrochloric acid (200 ml.)/ice (about 200 g.), and the precipitate was filtered off and washed with water. After drying, 22 g. of crude 4-n-hexylthiobenzoic acid with a melting point of 89—93° C. was obtained. A sample repeatedly

recrystallized from cyclohexane had a melting point of 96—98° C.

#### EXAMPLE 2

##### 4 - (4 - Phenylbutoxy) - benzoic acid

A solution of ethyl 4-hydroxy benzoate (11 g.), 4 - phenylbutylbromide (17 g.) and sodium (1.53 g.) in ethanol (50 ml.) was refluxed for 20 hours and was then evaporated *in vacuo*. 4 N sodium hydroxide (25 ml.) was added to the residue, and the mixture was heated on a steam bath for 5 hours. After cooling the resulting solution was acidified with concentrated hydrochloric acid (15 ml.). The precipitated material was collected by filtration and washed with water. After drying, 4 - (4 - phenylbutoxy) - benzoic acid with a melting point of 128—131° C. was obtained. Recrystallization twice from aqueous ethanol raised the melting point to 130—132° C. By substituting in the above procedure equimolar amounts of 2 - n - butylthioethylchloride for the 4 - phenylbutylbromide, 4 - (2 - n - butylthioethoxy) - benzoic acid (m.p. 95—97° C.), was obtained.

#### EXAMPLE 3

##### 4 - Piperidinomethyl - piperidine dihydrochloride hydrate

To a stirred mixture of piperidine (12 g.), potassium carbonate (28 g.) and methanol (100 ml.), 4 - chloromethylpiperidine hydrochloride (16.4 g.) was added in portions. The mixture was stirred at room temperature for a further 2 hours and was then evaporated *in vacuo*. The residue was treated with 2 N sodium hydroxide (25 ml.) and the separated oil was extracted with diethyl ether. The organic phase was dried (MgSO<sub>4</sub>) and distilled. 4 - Piperidinomethylpiperidine with a boiling point of 126—126.5° C. at 9 mm. Hg. was obtained. This material was dissolved in a mixture of methanol (75 ml.) and 3 N hydrochloric acid (45 ml.) and was hydrogenated after addition of PtO<sub>2</sub> (0.5 g.). The hydrogen uptake was complete within 3.5 hours. The catalyst was removed by filtration and the filtrate was evaporated *in vacuo*. The crystalline residue was triturated with acetone and collected by filtration. After drying, 4 - piperidinomethylpiperidine dihydrochloride hydrate with a melting point of about 260° C. was obtained. Recrystallization from ethanol raised the melting point to 265—266° C.

#### EXAMPLE 4

##### 4 - [2 - (4 - Methylpiperidino) - ethyl] - piperidine dihydrochloride

A mixture of 4-vinylpyridine (25 g.), 4 - methylpiperidine (35.4 g.) and acetic acid (3.5 ml.) was heated on a steam bath for 24 hours. 4 N Sodium hydroxide (25 ml.) was added to the cooled mixture and the separated oil was extracted with diethyl ether.

The organic phase was dried ( $\text{MgSO}_4$ ) and distilled. 4 - [2 - (4 - methylpiperidino) - ethyl] - pyridine with a boiling point of 151–154° C. at 9 mm. Hg. was obtained.

- 5 This material was dissolved in a mixture of methanol (230 ml.) and 4 N hydrochloric acid (130 ml.) and was hydrogenated after addition of  $\text{PtO}_2$  (1.0 g.). The hydrogen uptake was complete within 20 hours. The catalyst was removed by filtration and the filtrate was evaporated *in vacuo*. The crystalline residue was triturated with acetone and was collected by filtration. After drying, 4 - [2 - (4 - methylpiperidino) - ethyl] - piperidine dihydrochloride with a melting point higher than 290° C. was obtained. By substituting in the above procedure equimolar amounts of 3 - methylpiperidine or N-methylcyclohexylamine for the 4 - methylpiperidine, 4 - [2 - (3 - methylpiperidino) - ethyl] - piperidine dihydrochloride (m.p. 267–269° C.) and 4 - (2 - N - methylcyclohexylaminomethyl) - piperidine dihydrochloride (hygroscopic) were obtained respectively.

#### EXAMPLE 5

- 1 - (4 - n - hexyloxybenzoyl) - 4 - (2 - piperidinoethyl) - piperidine hydrochloride  
A solution of 4 - n - hexyloxybenzoyl chloride (5.0 g.) in methylenechloride (25 ml.) was slowly added to a mixture of 4 - (2 - piperidinoethyl) piperidine dihydrochloride (5.4 g.), methylenechloride (25 ml.) and 2 N sodium hydroxide (50 ml.) while stirring at 0–5° C. After the addition was complete the stirring was continued for a further 4 hours. The organic layer was separated, washed with brine, dried ( $\text{MgSO}_4$ ) and evaporated *in vacuo*. The remaining material was dissolved in diethylether (50 ml.) and acidified with dry ethanoic hydrochloric acid. The precipitated oily material was crystallized from isopropanol/diethylether. After drying and recrystallization from acetone, 6.3 g. of 1 - (4 - n - hexyloxybenzoyl) - 4 - (2 - piperidinoethyl) - piperidine hydrochloride with a melting point of 197.5–198.5° C was obtained.

#### EXAMPLE 6

- By substituting in the above procedure equimolar amounts of 4 - n - propyloxybenzoyl chloride, 4 - isopropoxybenzoyl chloride, 4 - n - butyloxybenzoyl chloride, 4 - sec - butyloxybenzoyl chloride, 4 - isobutyloxybenzoyl chloride, 4 - isoamylbenzoyl chloride, 4 - n - heptyloxybenzoyl chloride, 4 - n - octyloxybenzoyl chloride, 3 - n - propyloxybenzoyl chloride, 3 - n - butyloxybenzoyl chloride, 3 - n - amylbenzoyl chloride, 3 - n - hexyloxybenzoyl chloride, 4 - n - hexylthiobenzoate, 4 - n - pentylbenzoyl chloride, 4 - n - hexylbenzoyl chloride, 4 - (2 - phenylethoxy) - benzoate, 4 - (3 -

phenylpropoxy) - benzoate, 4 - (4 - phenyl butoxy) - benzoate, 4 - (2 - phenoxyethoxy) - benzoate, 4 - (2 - n - butylthioethoxy) - benzoate, 4 - n - heptylbenzoyl chloride, or 4 - n - octylbenzoyl chloride, for the 4 - n - hexyloxybenzoyl chloride, 1 - (4 - n - propyloxybenzoyl) - 4 - (2 - piperidinoethyl) - piperidine hydrochloride (m.p. 192–193.5° C.), 1 - (4 - isopropoxybenzoyl) - 4 - (2 - piperidinoethyl) - piperidine hydrochloride (m.p. 218–220° C.), 1 - (4 - n - butyloxybenzoyl) - 4 - (2 - piperidinoethyl) - piperidine hydrochloride (m.p. 194.5–195.5° C.), 1 - (4 - sec - butyloxybenzoyl) - 4 - (2 - piperidinoethyl) - piperidine hydrochloride (m.p. 172–174° C.), 1 - (4 - isobutyloxybenzoyl) - 4 - (2 - piperidinoethyl) - piperidine hydrochloride (m.p. 189–191° C.), 1 - (4 - isoamylbenzoyl) - 4 - (2 - piperidinoethyl) - piperidine hydrochloride (m.p. 207–209° C.), 1 - (4 - n - heptyloxybenzoyl) - 4 - (2 - piperidinoethyl) - piperidine hydrochloride (m.p. 197–199° C.), 1 - (4 - n - octyloxybenzoyl) - 4 - (2 - piperidinoethyl) - piperidine hydrochloride (m.p. 198.5–199.5° C.), 1 - (3 - n - propyloxybenzoyl) - 4 - (2 - piperidinoethyl) - piperidine hydrochloride semihydrate (m.p. 156.5–158.5° C.), 1 - (3 - n - butyloxybenzoyl) - 4 - (2 - piperidinoethyl) - piperidine hydrochloride (m.p. 135.5–136.5° C.), 1 - (3 - n - amylbenzoyl) - 4 - (2 - piperidinoethyl) - piperidine hydrochloride (m.p. 124–126° C.), 1 - (3 - n - hexyloxybenzoyl) - 4 - (2 - piperidinoethyl) - piperidine hydrochloride (m.p. 135–136° C.), 1 - (4 - n - hexylthiobenzoate) - 4 - (2 - piperidinoethyl) - piperidine hydrochloride (m.p. 166–168° C.), 1 - (4 - n - butylbenzoyl) - 4 - (2 - piperidinoethyl) - piperidine hydrochloride (m.p. 171.5–172.5° C.), 1 - (4 - n - pentylbenzoyl) - 4 - (2 - piperidinoethyl) - piperidine hydrochloride (m.p. 170.5–172° C.), 1 - (4 - n - hexylbenzoyl) - 4 - (2 - piperidinoethyl) - piperidine hydrochloride (m.p. 172.5–173.5° C.), 1 - [4 - (2 - phenylethyl) - benzoate] - 110  
1 - (4 - (2 - piperidinoethyl) - piperidine hydrochloride (m.p. 190–191.5° C.), 1 - [4 - (3 - phenyl propoxy) - benzoate] - 4 - (2 - piperidinoethyl) - piperidine hydrochloride (m.p. 204–205° C.), 1 - [4 - (4 - phenylbutoxy) - benzoate] - 4 - (2 - piperidinoethyl) - piperidine hydrochloride (m.p. 158.5–160° C.), 1 - [4 - (2 - phenoxyethoxy) - benzoate] - 4 - (2 - piperidinoethyl) - piperidine (m.p. 103–106° C.), 1 - [4 - (2 - n - butylthioethoxy) - benzoate] - 4 - (2 - piperidinoethyl) - piperidine hydrochloride (m.p. 161–163° C.), 1 - (4 - n - heptylbenzoyl) - 4 - (2 - piperidinoethyl) - piperidine hydrochloride (m.p. 173–173.5° C.), and 1 - (4 - n - octylbenzoyl) - 4 - (2 - piperidinoethyl) - piperidine hydrochloride (m.p. 177–179° C.), were obtained.

## EXAMPLE 7

- By substituting in the procedure described in the Example 5 above equimolar amounts of 4 - piperidinomethyl - piperidine dihydrochloride, 4 - (3 - piperidinopropyl) - piperidine dihydrochloride, 4 - (2 - N - methylcyclohexylaminoethyl)piperidine dihydrochloride, 4 - (2 - morpholinoethyl) - piperidine dihydrochloride, 3 - (piperidinomethyl) - piperidine dihydrochloride, 4 - (2 - pyrrolidinoethyl) - piperidine dihydrochloride, 4 - (2 - hexamethyleneiminoethyl) - piperidine dihydrochloride, 4 - [2 - (3 - methylpiperidino) - ethyl] - piperidine dihydrochloride, or 4 - [2 - (4 - methylpiperidino) - ethyl] - piperidine dihydrochloride, for the 4 - (2 - piperidinoethyl) - piperidine dihydrochloride, 1 - (4 - n - hexyloxybenzoyl) - 4 - piperidinomethyl piperidine hydrochloride (m.p. 199—200.5° C.), 1 - (4 - n - hexyloxybenzoyl) - 4 - (3 - piperidinopropyl) - piperidine hydrochloride (m.p. 147—149° C.), 1 - (4 - n - hexyloxybenzoyl) - 4 - (2 - N - methylcyclohexylaminoethyl) - piperidine hydrochloride (m.p. 194—195° C.), 1 - (4 - n - hexyloxybenzoyl) - 4 - (2 - morpholinoethyl) - piperidine hydrochloride (m.p. 185.5—187° C.), 1 - (4 - n - hexyloxybenzoyl) - 3 - (piperidinomethyl) - piperidine hydrochloride (m.p. 149—151° C.), 1 - (4 - n - hexyloxybenzoyl) - 4 - (2 - pyrrolidinoethyl) - piperidine hydrochloride (m.p. 142.5—144.5° C.), 1 - (4 - n - hexyloxybenzoyl) - 4 - (2 - hexamethyleneiminoethyl) - piperidine hydrochloride (m.p. 166.5—168.5° C.), 1 - (4 - n - hexyloxybenzoyl) - 4 - [2 - (3 - methylpiperidino) - ethyl] - piperidine hydrochloride (m.p. 174.5—177° C.), and 1 - (4 - n - hexyloxybenzoyl) - 4 - [2 - (4 - methyl - piperidino) - ethyl] -

piperidine hydrochloride (m.p. 178—180° C.) were obtained.

## EXAMPLE 8

By using in the procedure described in the Example 5 above and substituting equimolar amounts of 4 - n - amyloxybenzoyl chloride for the 4 - n - hexyloxybenzoyl chloride and 48% hydrobromic acid for the ethanolic hydrochloric acid, 1 - (4 - n - amyloxybenzoyl) - 4 - (2 - piperidinoethyl) - piperidine hydrobromide (m.p. 192.5—194° C.) was obtained.

## EXAMPLE 9

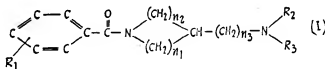
Tablets containing 5 mg. of the active compound, and the following composition, were prepared as follows:

1 - (4 - n - hexyloxybenzoyl) - 4 - (piperidinoethyl) - piperidine, hydrochloride	5 mg.
Lactose	69 mg.
Starch	57 mg.
Gelatin	2 mg.
Talc	9 mg.

The calculated amount of active compound, lactose, and starch, were mixed, granulated with a solution of gelatine in water and dried. After sifting the calculated amount of talc is added, the tablets are made by means of a 17 mm. punching die which provides tablets weighing 142 mg. each corresponding to 5 mg. of the active compound per tablet.

## WHAT WE CLAIM IS:—

1. A compound of the general formula



wherein  $\text{R}_1$  is a straight or branched  $\text{C}_3$  to  $\text{C}_{12}$  aliphatic hydrocarbon chain, unsubstituted or substituted with phenyl, phenoxy or phenylthio, which  $\text{R}_1$  is directly attached to the benzene nucleus, or optionally may be attached to the benzene nucleus through a hetero atom which is an oxygen or sulphur atom;  $\text{R}_2$  is alkyl;  $\text{R}_3$  is a cycloalkyl radical with from 5 to 8 carbon atoms in the ring, and  $\text{R}_2$  together with  $\text{R}_3$  can complete a heterocyclic ring which may be alkyl-substituted,  $n_1$  is an integer from 2 to 4,  $n_2$  is an integer from 1 to 5, and  $n_3$  is an integer from 1 to 3; or a salt of the compound with a pharmaceutically acceptable inorganic or organic acid.

2. A compound according to Claim 1 in

which  $n_2$  is 2 or 3, and  $\text{R}_1$  is attached to the nucleus at the 4-position.

3. A compound according to Claims 1 and 2 in which  $n_1$  and  $n_2$  have the value 2, and  $\text{R}_1$  is a  $\text{C}_5$  to  $\text{C}_7$  aliphatic hydrocarbon chain attached to the benzene nucleus through the hetero atom O.

4. A compound according to claims 1, 2 and 3 in which  $\text{R}_2$  and  $\text{R}_3$  form part of a heterocyclic ring.

5. A compound as claimed in claim 4, in which  $\text{R}_2$  and  $\text{R}_3$  form part of a heterocyclic ring selected from the group consisting of the pyrrolidino, the piperidino, the hexamethylenimino, and the heptamethylenimino rings.

6. A compound as claimed in claim 5 in



one of Examples 1 to 8 of the foregoing  
Examples.

22. A pharmaceutical preparation in dos-  
age unit form substantially as hereinbefore  
described in Example 9 of the foregoing  
Examples.

TREGEAR, THIEMANN & BLEACH,  
Chartered Patent Agents,  
Melbourne House, Aldwych,  
London, W.C.2.  
Agents for the Applicant(s).

Printed for Her Majesty's Stationery Office, by the Courier Press, Leamington Spa, 1971.  
Published by The Patent Office, 25 Southampton Buildings, London, WC2A 1AY, from  
which copies may be obtained.